

PCTWORLD INTELLECTUAL PROPERTY ORGANIZATION
International Bureau

INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification ⁶ : A61K 7/16	A1	(11) International Publication Number: WO 99/44570 (43) International Publication Date: 10 September 1999 (10.09.99)
(21) International Application Number: PCT/EP98/02963 (22) International Filing Date: 20 May 1998 (20.05.98) (30) Priority Data: FI98A000051 6 March 1998 (06.03.98) IT (71) Applicant (for all designated States except US): ITALMED S.N.C. DI GALLI G. E PACINI G. [IT/IT]; Via Andrea del Castagno, 12, I-50132 Firenze (IT). (72) Inventor; and (75) Inventor/Applicant (for US only): GALLI, Giovanna [IT/IT]; Via Giambologna, 39, I-50132 Firenze (IT). (74) Agent: CELESTINO, Marco; ABM Agenzia Brevetti & Marchi, Via A. Della Spina, 40, I-56125 Pisa (IT).		(81) Designated States: AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CU, CZ, DE, DK, EE, ES, FI, GB, GE, GH, GM, GW, HU, ID, IL, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, UA, UG, US, UZ, VN, YU, ARIPO patent (GH, GM, KE, LS, MW, SD, SZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, ML, MR, NE, SN, TD, TG). Published <i>With international search report.</i>
(54) Title: DESENSITIZING DENTAL COMPOSITION (57) Abstract A dental composition for the treatment of dentinal hypersensitivity and in particular for the desensitization of exposed dentin, the desensitizing treatment of deep cavities, the desensitizing treatment when replacing dental layers, the stumps desensitizing treatment before placing dental prosthesis. The composition can be used as a solution or as a gel. In the first case two distinct liquid solutions for use successively on the exposed dentin are provided for. In the second case two distinct gel compounds spread successively on the exposed dentin are provided for. The first solution or the first gel compound comprises preferably three soluble potassium salts, whereas the second solution or the second gel compound comprises a calcium salt and a soluble strontium salt. In a preferred composition two solutions are provided for of which the first has solutes comprising potassium phosphate, potassium carbonate and potassium fluoride, and the second solution has solutes comprising calcium chloride and strontium chloride. For the gel composition, which may be used as toothpaste, two distinct gel compounds are provided, the first having solutes comprising potassium phosphate, potassium carbonate and potassium fluoride and the second gel compound having solutes comprising calcium chloride and strontium chloride.		

FOR THE PURPOSES OF INFORMATION ONLY

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AL	Albania	ES	Spain	LS	Lesotho	SI	Slovenia
AM	Armenia	FI	Finland	LT	Lithuania	SK	Slovakia
AT	Austria	FR	France	LU	Luxembourg	SN	Senegal
AU	Australia	GA	Gabon	LV	Latvia	SZ	Swaziland
AZ	Azerbaijan	GB	United Kingdom	MC	Monaco	TD	Chad
BA	Bosnia and Herzegovina	GE	Georgia	MD	Republic of Moldova	TG	Togo
BB	Barbados	GH	Ghana	MG	Madagascar	TJ	Tajikistan
BE	Belgium	GN	Guinea	MK	The former Yugoslav Republic of Macedonia	TM	Turkmenistan
BF	Burkina Faso	GR	Greece	ML	Mali	TR	Turkey
BG	Bulgaria	HU	Hungary	MN	Mongolia	TT	Trinidad and Tobago
BJ	Benin	IE	Ireland	MR	Mauritania	UA	Ukraine
BR	Brazil	IL	Israel	MW	Malawi	UG	Uganda
BY	Belarus	IS	Iceland	MX	Mexico	US	United States of America
CA	Canada	IT	Italy	NE	Niger	UZ	Uzbekistan
CF	Central African Republic	JP	Japan	NL	Netherlands	VN	Viet Nam
CG	Congo	KE	Kenya	NO	Norway	YU	Yugoslavia
CH	Switzerland	KG	Kyrgyzstan	NZ	New Zealand	ZW	Zimbabwe
CI	Côte d'Ivoire	KP	Democratic People's Republic of Korea	PL	Poland		
CM	Cameroon	KR	Republic of Korea	PT	Portugal		
CN	China	KZ	Kazakhstan	RO	Romania		
CU	Cuba	LC	Saint Lucia	RU	Russian Federation		
CZ	Czech Republic	LI	Liechtenstein	SD	Sudan		
DE	Germany	LK	Sri Lanka	SE	Sweden		
DK	Denmark	LR	Liberia	SG	Singapore		
EE	Estonia						

- 1 -

TITLE

DESENSITIZING DENTAL COMPOSITION

DESCRIPTION5 Field of the invention

The present invention generally relates to a compound for dentistry and, more precisely, it relates to a dental composition for the treatment of dentinal hypersensitivity.

10 In particular, the composition is suitable for the desensitization of the exposed dentin, such as, in particular, for

- the desensitizing treatment of deep cavities,
- the desensitizing treatment when replacing dental
- 15 layers,
- the stumps desensitizing treatment before placing dental prosthesis.

The composition can be used as a solution or as a gel.

20 Description of the prior art

Dentinal hypersensitivity is a problem often met by dentists in their patients and is often intensified when eating hot or cold food, sweet or acidulous food, as well as when brushing teeth.

25 Normal dentin is covered by enamel (crown dentin) or by cement (radicular dentin) and is not permeable.

In teeth sensitive zones enamel is found often eroded or worn away and dentin exposed. In other cases, gingival recession uncovers the most sensitive portions

30 of the teeth, i.e. the junction enamel-cement.

The absence of cover exposes the dentinal tubules. In addition, pulp is rich of nerves many of which are centrifugally directed towards dentin. Dentin is crossed

CONFIRMATION COPY

- 2 -

radially by dentinal tubules which contain dentinal fluid. Through the tubules a nervous stimulus reaches sensorial areas of the dental pulp.

A first known way to reduce dentinal sensitivity is to close the orifices of dentinal tubules. To this extent, natural substances are known able to reduce sensitivity by closing the tubules, such as tartar, collagen or mineral salts precipitated in saliva.

Chemical compounds, like solutions, gel compounds or pastes, as well as physical treatments are also known capable of mechanically obliterating the orifices of the dentinal tubules.

Some of these compounds or treatments are:

- potassium and iron oxalate, which react with ionised calcium present in the saliva and form calcium oxalate which precipitates and obliterates the dentinal tubules orifices;
- silver nitrate which acts through the precipitation of silver compounds;
- tin fluoride, which acts through the precipitation of tin compounds;
- strontium salts;
- insoluble salts such as calcium phosphate, hydroxyapatite, colloidal silica etc. which directly obliterate the tubular orifices;
- glass-ionomeric cement;
- some types of resins.

In any case the obliteration of the dentinal tubules by means of soluble or insoluble salts used up to now, even if they are simple to use, is not enough effective and last only a short time.

Compounds are known, moreover, such as potassium chloride or nitrate or oxalate capable of reducing the dentinal sensitivity through a depolarising effect of the

- 3 -

nervous fibres, without obliterating the dentinal tubules. Therefore, desensitization treatments are known using said compounds, capable of stopping the pulpal nervous activity by varying the dentinal tubules nervous fibres excitability. However, also in this case, the effect duration is short.

New techniques, not yet used in clinical practice, use glass-ionomeric cement or photo-induced resins. Desensitization is improved, but they are not easy to use and are good for deep erosions only.

Finally, treatments are known using either laser or ultrasounds scalers, which cause the formation of smear layers and tubules obliteration. These treatments give results, such as effectiveness and duration, similar to the topical treatment with the above described soluble or insoluble salts, but they require expensive apparatus.

Summary of the invention

It is an object of the present invention to provide a dental composition for dentinal desensitization which has good mechanical obliteration properties of the dentinal tubules in addition to depolarising properties of the nervous dentinal fibres, and is capable of obtaining good effectiveness and long duration, as well as of being easy to use.

According to a first aspect of the invention, this object is reached by the composition whose characteristic is to comprise at least two solutions to be used successively on the exposed dentin.

A second aspect of the invention provides at least two distinct gel compounds to be used successively on the exposed dentin.

The characteristics of said solutions and said gel compounds is to comprise a first and a second solution or a first and a second gel, suitable for being mixed

- 4 -

topically, wherein:

- in the first solution a first solute consisting in potassium phosphate and at least a second solute selected among a potassium salt are present, and
- 5 - in the second solution a first solute selected among a calcium salt and at least a second solute selected among a strontium salt, a silver salt, a barium salt, a zinc salt are present.

The preferred general weight ratio for the first
10 solution is the following:

	potassium phosphate	0,5-20%
	potassium carbonate	0,5-10%
	potassium fluoride	0,1-5%
	sodium methylparaben	0,1%
15	deionized water	64,9-98,8%

The preferred general weight ratio for the second
solution is the following:

	calcium chloride	0,5-20%
	strontium chloride	0,5-11%
20	benzoate sodium	0,2%
	deionized water	68,8-98,8%

The preferred general weight ratio for the first
gel compound is the following:

	potassium phosphate	0,5-10%
25	potassium carbonate	0,5-5%
	potassium fluoride	0,1-0,5%
	sorbitol	30-45%
	colloidal silica	15-30%
	glycerol	5-10%
30	carboxy methyl hydroxy ethyl cellulose	0,5-1,5%
	lauryl sulphate sodium	1-1,5%
	benzoate sodium	0,3-0,8%
	saccharinated sodium	0,3-0,5%
	mint fragrance	q.s.

- 5 -

colour CI 42051, CI 19140	q.s.
purified water	q.s. 100 ml

The preferred general weight ratio for the second gel compound is the following:

5	strontium chloride	0,5-10%
	calcium chloride	0,5-10%
	sorbitol	30-45%
	colloidal silica	15-30%
	glycerol	5-10%
10	carboxy methyl hydroxy ethyl cellulose	0,5-1,5%
	lauryl sulphate sodium	1-1,5%
	benzoate sodium	0,3-0,8%
	saccharinated sodium	0,3-0,5%
	mint fragrance	q.s.
15	colour CI 16255, CI 47005	q.s.
	purified water	q.s. 100 ml

The combined use of the two preferred solutions or of the two preferred gel compounds successively has the result, after an immediate double exchange reaction, of six insoluble salts:

- calcium phosphate,
- calcium carbonate,
- calcium fluoride,
- strontium phosphate,
- 25 - strontium carbonate,
- strontium fluoride,

and a soluble salt, i.e. potassium chloride.

According to a sudden reaction a "crystal complex" is formed by said six insoluble salts which surprisingly have shown desensitizing properties in the short term (15 minutes) and in the long term (6-12 months).

The formation of the potassium chloride helps to increase the desensitization properties adding a nervous

- 6 -

depolarising effect. In fact, potassium chloride is obtained after a double exchange reaction and is present in solution inside the dentinal tubule when the "crystal complex" is formed which obliterates the tubules orifices. In this way a higher amount of potassium is present for a longer time with respect to the depolarising compounds according to the prior art, which do not obliterate the dentinal tubules, thus allowing the dentinal fluid (whose flow is always centrifugal) and then also the potassium ions to come out the dentinal tubules.

The use of an induced crystallisation for obliterating the dentinal tubules with the formation of a "crystal complex" along with the formation (always inside the tubules) of a compound with depolarising properties of the nervous fibres, is new in the dental field, and is used with excellent results according to the present invention.

If the composition according to the invention is prepared with potassium phosphate, potassium carbonate and potassium fluoride (for the first solution) in amounts higher than 20, 10 and 5% respectively for the first solution and with calcium chloride and strontium chloride in amounts higher than 20 and 11% respectively for the second solution, a desensitizing is obtained having the same good properties as above described, but less convenient to use. In fact, at the moment of the union of the two solutions, when the formation of the six insoluble salts occurs, an opalescent gel compound would result, less handy and less spreadable on the surface to cure.

It is also advisable that the two solutions are spread with different brushes so that the two solutions or gels are not in contact before they reach the patient's mouth.

- 7 -

Similarly, the composition with potassium phosphate, potassium carbonate and potassium fluoride in amounts higher than 10, 5 and 0,5% for the first gel compound and with calcium chloride and strontium chloride in amounts higher than 10% for both the salts of the second gel compound, a desensitizing composition is obtained which can be used having the same properties as above described, but less suitable for a toothpaste because its organoleptic aspect would be less acceptable.

The composition according to the present invention, will be made clearer with the not limitative following examples.

EXAMPLE 1.1

A composition for desensitizing exposed dentin which uses two solutions to be spread successively comprises potassium phosphate, potassium carbonate and potassium fluoride for the first solution and calcium chloride and strontium chloride for the second solution, with the following weight ratio:

20	Solution n° 1	
	potassium phosphate	16%
	potassium carbonate	5%
	potassium fluoride	3%
	sodium methylparaben	0.1%
25	deionized water	75,9%
	Solution n° 2	
	calcium chloride	16%
	strontium chloride	10,6%
	benzoate sodium	0.2%
30	deionized water	73,2%

The preferred spreading method is the following for the desensitizing treatment: after having isolated the surgical area with cotton elements or the like and after having ablated the dental plaque by means of low speed

- 8 -

electric brushing, the surface to be treated is cleaned by cotton pellets wet by a disinfectant liquid (such as sodium hypochlorite 5% solution). Then the surface is dried with air jet for about 15 seconds. The drying
5 action must substantially dehydrate the external layer of the exposed dentin. Then, by means of a brush or cotton pellets or spongy elements gently rubbed, the first solution is spread for about 20 seconds. Immediately after the second solution is spread onto the same dental
10 surface and in the same way.

When hypersensitivity is high, the same treatment can be repeated.

The action of such a composition, which reacts forming a crystal complex which deeply obliterates the
15 dentinal tubules orifices is double. In fact, the first solution spread on the dehydrated dentin causes, for capillarity, the filling of the dentinal tubules. To this the nervous depolarisation is added of the potassium chloride which always forms inside the tubules through a
20 double exchange reaction.

Alternatively to the composition of the above example 1, in the same way the following exemplifying compositions can be used. In these compositions only two potassium salts are present in the first solution and the
25 crystal complex will be formed by 4 insoluble salts only. Soluble potassium salts are still present.

EXAMPLE 1.2

Solution n° 1

	potassium phosphate	16%
30	potassium oxalate	6%
	sodium methylparaben	0.1%
	deionized water	q.s. 100 ml

Solution n° 2

	calcium chloride	16%
--	------------------	-----

- 9 -

strontium chloride	10%
benzoate sodium	0.2%
deionized water	73,8%

EXAMPLE 1.3

5	Solution n° 1	
	potassium phosphate	16%
	potassium carbonate	5%
	sodium methylparaben	0.1%
	deionized water	q.s. 100 ml

10	Solution n° 2	
	calcium chloride	16%
	barium chloride	10%
	benzoate sodium	0.2%
	deionized water	73,8%

EXAMPLE 1.4

15	Solution n° 1	
	potassium phosphate	16%
	potassium carbonate	5%
	sodium methylparaben	0.1%
20	deionized water	q.s. 100 ml

	Solution n° 2	
	calcium chloride	16%
	silver chloride	10%
	benzoate sodium	0.2%
25	deionized water	73,8%

EXAMPLE 1.5

	Solution n° 1	
	potassium phosphate	16%
	potassium carbonate	5%
30	sodium methylparaben	0.1%
	deionized water	q.s. 100 ml

Solution n° 2

- 10 -

calcium chloride	16%
zinc chloride	10%
benzoate sodium	0.2%
deionized water	73.8%

5

EXAMPLE 1.6

Solution n° 1

potassium phosphate	16%
potassium carbonate	5%
10 sodium methylparaben	0.1%
deionized water	78,9%

Solution n° 2

calcium chloride	16%
strontium chloride	10,6%
15 benzoate sodium	0.2%
deionized water	73,2%

EXAMPLE 2

A desensitizing composition to be used as a
20 toothpaste which uses sequentially two gel compounds
comprising potassium phosphate, potassium carbonate and
potassium fluoride for the first gel compound and calcium
chloride and strontium chloride for the second gel
compound, with the following weight ratio:

25 Gel compound n°1

potassium phosphate	8%
potassium carbonate	3,5%
potassium fluoride	0,4%
sorbitol	30%
30 colloidal silica	15%
glycerol	5%
lauryl sulphate sodium	1,5%
carboxy methyl hydroxy ethyl cellulose	1%

- 11 -

	benzoate sodium	0,5%
	saccharinated sodium	0,4%
	mint fragrance	q.s.
	colour CI 42051, CI 19140	q.s.
5	purified water	q.s. 100 ml
	Gel compound n°2	
	calcium chloride	7%
	strontium chloride	6%
	sorbitol	30%
10	colloidal silica	15%
	glycerol	5%
	lauryl sulphate sodium	1,5%
	carboxy methyl hydroxy ethyl cellulose	1%
	benzoate sodium	0,5%
15	saccharinated sodium	0,4%
	mint fragrance	q.s.
	colour CI 16255, CI 47005	q.s.
	purified water	q.s. 100 ml

This composition is used in the following way for
 20 the desensitizing treatment:

- on the toothbrush an amount of gel compound n°1 substantially equal to the volume of two peas is poured;
- both dental arches are spread with gel compound n°1 by brushing from the above to the bottom for about two
 25 minutes;
- the friction is prolonged on the areas which are sensitive to hot or cold food, to acidulous or sweet substances;
- without rinsing, after having spread on the toothbrush
 30 the same amount of gel compound n°2, the same dental surface is brushed again as above described for the first gel compound;
- after the two gel compounds have mixed on the dental surface an instant double exchange reaction occurs with

- 12 -

the formation of the "crystal complex", comprising the six insoluble salts and potassium chloride;

- then the mouth is rinsed with water.

The foregoing description of specific embodiments
5 will so fully reveal the invention according to the conceptual point of view, so that others, by applying current knowledge, will be able to modify and/or adapt for various applications such embodiments without further research and without departing from the invention, and it
10 is therefore to be understood that such adaptations and modifications will have to be considered as equivalent to the specific embodiments. The means and the materials to realise the different functions described herein could have a different nature without, for this reason,
15 departing from the field of the invention. It is to be understood that the phraseology or terminology employed herein is for the purpose of description and not of limitation.

- 13 -

CLAIMS

1. A dental composition for the desensitization of the exposed dentin characterised in that it comprises a first and a second liquid or gel solution, suitable for being
5 mixed topically, wherein:
- in said first solution a first solute consisting in potassium phosphate and at least a second solute selected among a potassium salt are present, and
 - in said second solution a first solute selected among a
10 calcium salt and at least a second solute selected among a strontium salt, a silver salt, a barium salt, a zinc salt are present.
2. Composition according to claim 1, wherein
- in said first liquid solution said first solute is
15 potassium phosphate said at least second solute is selected among potassium carbonate, potassium fluoride, potassium oxalate, and
 - in said second liquid solution the first solute is selected among calcium chloride or calcium acetate and the
20 second solute is selected among strontium chloride, strontium acetate, silver chloride, barium chloride, zinc chloride.
3. Composition according to claim 1, wherein
- in said first gel compound said first solute is
25 potassium phosphate and said at least second solute is selected among potassium carbonate and potassium fluoride, and
 - in said second gel compound the first solute is calcium chloride and the second solute is strontium chloride.
- 30 4. Composition according to claims 1 or 2, characterised in that it comprises a solvent and a bacteriostatic preservative compound.
5. Composition according to claim 4 wherein the solutes are potassium phosphate, potassium carbonate and potassium

fluoride for the first solution; calcium chloride and strontium chloride for the second solution, and are mixed to said solvent and to said preservative compound according to the following general weight ratio:

5 Solution n°1

potassium phosphate	0,5-20%
potassium carbonate	0,5-10%
potassium fluoride	0,5-5%
Preservative	0,1%
Solvent	64,9-98,4%

10

Solution n°2

calcium chloride	0,5-20%
strontium chloride	0,5-10%
Preservative	0,2%
Solvent	69,8-98,8%

15

6. Composition according to claims 4 or 5 wherein said solvent is deionized water and said preservative compound is sodium methylparaben for the solution 1 and benzoate sodium for the solution 2.

20 7. Composition according to claim 6 wherein said solutes, said solvent, and said preservative compound are mixed in the following weight ratio.

Solution n°1

potassium phosphate	16%
potassium carbonate	5%
potassium fluoride	3%
sodium methylparaben	0,1%
deionized water	75,9%

25

Solution n°2

calcium chloride	15%
strontium chloride	10,6%
benzoate sodium	0,2%
deionized water	74,2%

30

8. Composition such as to claim 4, wherein

- 15 -

- in said first liquid solution said at least second solute is potassium oxalate, and
 - in said second liquid solution the first solute is selected among calcium chloride or acetate and the second solute is selected among strontium chloride or acetate, silver chloride, barium chloride, zinc chloride, and
 - said solvent is deionized water and said preservative is sodium methylparaben for the first solution and benzoate sodium for the second solution.
9. Composition according to claim 3 for use as gel toothpaste capable of desensitizing the exposed dentin, wherein the solutes potassium phosphate, potassium carbonate and potassium fluoride for the first gel compound and the calcium chloride and the strontium chloride for the second gel compound and the other compounds according to the following general weight ratio:

	Gel compound n°1	
	potassium phosphate	0,5-10%
20	potassium carbonate	0,5-5%
	potassium fluoride	0,1-0,5%
	sorbitol	30-45%
	colloidal silica	15-30%
	glycerol	5-10%
25	carboxy methyl hydroxy ethyl cellulose	0,5-1,5%
	lauryl sulphate sodium	1-1,5%
	benzoate sodium	0,3-0,8%
	saccharinated sodium	0,3-0,5%
	mint fragrance	q.s.
30	colour CI 42051, CI 19140	q.s.
	purified water	q.s. 100 ml
	Gel compound n°2	
	strontium chloride	0,5-10%
	calcium chloride	0,5-10%

- 16 -

	sorbitol	30-45%
	colloidal silica	15-30%
	glycerol	5-10%
	carboxy methyl hydroxy ethyl cellulose	0,5-1,5%
5	lauryl sulphate sodium	1-1,5%
	benzoate sodium	0,3-0,8%
	saccharinated sodium	0,3-0,5%
	mint fragrance	q.s.
	colour CI 16255, CI 47005	q.s.
10	purified water	q.s. 100 ml

10. Composition according to claim 8 wherein said solutes and the other compounds are mixed in the following weight ratio:

	Gel compound n°1	
15	potassium phosphate	8%
	potassium carbonate	3,5%
	potassium fluoride	0,4%
	sorbitol	30%
	colloidal silica	15%
20	glycerol	5%
	lauryl sulphate sodium	1,5%
	carboxy methyl hydroxy ethyl cellulose	1%
	benzoate sodium	0,5%
	saccharinated sodium	0,4%
25	mint fragrance	q.s.
	colour CI 42051, CI 19140	q.s.
	purified water	q.s. 100 ml
	Gel compound n°2	
	calcium chloride	7%
30	strontium chloride	6%
	sorbitol	30%
	colloidal silica	15%
	glycerol	5%
	lauryl sulphate sodium	1,5%

- 17 -

	carboxy methyl hydroxy ethyl cellulose	1%
	benzoate sodium	0,5%
	saccharinated sodium	0,4%
	mint fragrance	q.s.
5	colour CI 16255, CI 47005	q.s.
	purified water	q.s. 100 ml

11. A method for making a dental desensitizing composition characterised in that it comprises the steps of:

- preparing a first liquid solution or gel solution
- 10 comprising a first solute consisting in potassium phosphate and at least a second solute selected among other potassium salts, and
- preparing a second liquid solution or gel solution comprising a first solute selected among a chloride or an
- 15 acetate salt;
- mixing topically the first and second solution thus forming: a crystal complex comprising a plurality of insoluble salts obtained by double exchange reaction of the above stated potassium, chloride or acetate salts,
- 20 said insoluble salts having obliterating properties of the dentinal tubules; and a soluble potassium salt which has depolarising effect on dentin.

INTERNATIONAL SEARCH REPORT

Inter: National Application No

PCT/EP 98/02963

A. CLASSIFICATION OF SUBJECT MATTER

IPC 6 A61K7/16

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 6 A61K

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	WO 92 04006 A (NEIRINCKX) 19 March 1992 see the whole document ---	1-11
A	US 5 603 922 A (WINSTON ET AL.) 18 February 1997 see column 3, line 36 - column 4, line 4 see column 4, line 44 - line 49 see column 6, line 56 - column 8, line 38; claims 1-18; examples 1-17 ---	1-11
A	WO 97 06774 A (ENAMELON) 27 February 1997 see the whole document ---	1-11
A, P	WO 98 13012 A (ENAMELON) 2 April 1998 see the whole document ---	1-11
	--- -/--	

☒ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

* Special categories of cited documents :

- "A" document defining the general state of the art which is not considered to be of particular relevance
- "E" earlier document but published on or after the international filing date
- "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- "O" document referring to an oral disclosure, use, exhibition or other means
- "P" document published prior to the international filing date but later than the priority date claimed

- "T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
- "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
- "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.
- "&" document member of the same patent family

Date of the actual completion of the international search

13 November 1998

Date of mailing of the international search report

20/11/1998

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2
NL - 2280 HV Rijswijk
Tel. (+31-70) 340-2040, Tx. 31 651 epo nl.
Fax: (+31-70) 340-3016

Authorized officer

Fischer, J.P.

INTERNATIONAL SEARCH REPORT

International Application No

PCT/EP 98/02963

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	GB 2 239 601 A (G-C SHIKA KOGYO K K) 10 July 1991 see the whole document ----	1-11
A,P	WO 98 11866 A (JOHANSEN ET AL.) 26 March 1998 see the whole document ----	1-11
A	WO 96 25914 A (AMERICAN DENTAL ASSOCIATION HEALTH FOUNDATION) 29 August 1996 see the whole document -----	1-11

INTERNATIONAL SEARCH REPORT

information on patent family members

International Application No

PCT/EP 98/02963

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
WO 9204006	A	19-03-1992	AU 8406991 A	30-03-1992
US 5603922	A	18-02-1997	AU 6682996 A	12-03-1997
			CA 2224274 A	27-02-1997
			CN 1192672 A	09-09-1998
			EP 0845976 A	10-06-1998
			WO 9706774 A	27-02-1997
WO 9706774	A	27-02-1997	US 5645853 A	08-07-1997
			US 5603922 A	18-02-1997
			AU 6682996 A	12-03-1997
			CA 2224274 A	27-02-1997
			CN 1192672 A	09-09-1998
			EP 0845976 A	10-06-1998
WO 9813012	A	02-04-1998	AU 4158297 A	17-04-1998
GB 2239601	A	10-07-1991	AU 6819090 A	04-07-1991
			BE 1003526 A	14-04-1992
			CA 2032556 C	24-10-1995
			CA 2032556 A	29-06-1991
			CH 682215 A	13-08-1993
			DE 4041946 A	18-07-1991
			FR 2656522 A	05-07-1991
			JP 4217904 A	07-08-1992
			SE 506444 C	15-12-1997
			SE 9004154 A	05-07-1991
			US 5234971 A	10-08-1993
WO 9811866	A	26-03-1998	AU 4342497 A	14-04-1998
			GB 2318977 A	13-05-1998
WO 9625914	A	29-08-1996	AU 695356 B	13-08-1998
			AU 1931995 A	11-09-1996
			EP 0871427 A	21-10-1998